

# Quality Characterization of Groundwater in some parts of Jajpur District, Odisha with Special reference to Drinking Water and Irrigation standards

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## ABSTRACT

Groundwater is one of the most significant resources impacting human health in terms of drinking water quality. The purpose of this study is to analyse the quality of groundwater of different sources and compare it with World Health Organization (WHO) standards and Bureau of Indian Standards (BIS). Groundwater samples were collected from several places across the study area. During 2018 pre-monsoon season, twenty samples were taken from the tube wells. The samples were analysed for determining their physico chemical characteristics in order to identify quality problems and recommend safe drinking water point sources. In addition to determination of major ions such as Sodium ( $\text{Na}^+$ ), Potassium ( $\text{K}^+$ ), Magnesium ( $\text{Mg}^{2+}$ ), Calcium ( $\text{Ca}^{2+}$ ), Sulphate ( $\text{SO}_4^{2-}$ ), Chloride ( $\text{Cl}^-$ ), Carbonate ( $\text{CO}_3^{2-}$ ), and Bicarbonate ( $\text{HCO}_3^-$ ), the suitability was determined by estimating Electrical conductivity (EC),  $\text{H}^+$  ion concentration (pH), Alkalinity, and Total dissolved Solid (TDS). The analysis results carried out showed the following concentration ranges pH (6.8-7.81), EC (103-1985  $\mu\text{s}/\text{cm}$ ), TH (133-466 mg/l), TDS (77-1469 mg/l), Ca (31-127 mg/l), Mg (9.97-436.2 mg/l), Na (10.3-88.3 mg/l), K (0.68-59 mg/l),  $\text{HCO}_3^-$  (96.5-436.2 mg/l),  $\text{Cl}^-$  (23.59-127.8 mg/l),  $\text{SO}_4^{2-}$  (15-57 mg/l), and alkalinity (92-289 mg/l). The Water Quality Index (WQI) technique is used to classify water quality in a region and ascertain if it is suitable for drinking. Out of 20 groundwater samples, 13 samples represented good quality water, 6 samples indicate poor quality water, and 1 sample indicated water to be unfit for consumptions for drinking purpose. Based on the physico chemical analyses results, different parameters are calculated for irrigation purposes like Percentage Sodium, Sodium Absorption Ratio, Salinity Index, Soluble sodium percentage, Residue Sodium Carbonate, Kelley's ratio and Magnesium Hazard Ratio. From the results it is found that most of the water samples are categorised under suitable condition for drinking and irrigation purposes and few samples are unsuitable for drinking and irrigation purposes.

**Key words :** *Physico chemical Study, WQI, Groundwater quality*

## Introduction

Out of the total fresh water reservoir available on earth, nearly 97% of all freshwaters (excluding glaciers and ice caps) is ground water. The remaining 3% comprises mainly of surface water which occur in lakes, rivers, wetlands, and soil moisture. Groundwater serves as major source of drinking

water across the world. Groundwater mainly occurs as sub surface water which moves slowly through aquifer unlike the surface water of the stream. So there is enormous chance of contamination as it comes in contact with the surrounding rock or sediment. Groundwater also gets contaminated by various man-made anthropogenic activities at the surface (or at depth). Besides the drinking purpose,

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Groundwater is also used for irrigation purpose to grow crops. In this study, groundwater is analysed for its suitability in drinking purpose and irrigation as well.

### Study Area

The study area (Fig. 1) lies between  $20^{\circ}34'57''$  and  $21^{\circ}08'52''$  N latitude and  $85^{\circ}41'21''$  and  $86^{\circ}37'26''$  E longitudes and in the survey of India toposheets they are exposed in 73L/5 and 73L/6. The district is bounded by Dhenkanal district on its east, Kendrapada district on its west, Keonjhar and Bhadrak districts on its North, and Cuttack on its south. The Geographical area of the district is 2888 square kilometer. A hot and humid climate is observed during April and May months, whereas in December and January, the climate remains cold. The Brahmani, the Baitarani, their tributaries and distributaries form the main drainage system in the study area. Good fertile soil is observed in the study

area. Geologically the district is essentially covered in the northwestern and southwestern parts under hilly areas comprising the Metasediments of Gorumahisani Group and Eastern Ghat Super group with basic and ultramafic intrusion. The different geomorphic features in the study area are hills, pediments, peneplains, valleys and the structural features constitute lineaments, faults, and joint and parallel drainage pattern (District Survey Report of Jajpur district Odisha, 2018).

### Methodology

Altogether 20 ground water samples were collected from a part of Jajpur district during the pre-monsoon season in June 2018. One-liter pre-washed polyethylene bottles with tightly fitting covers are used for the collection of samples. At the time of sampling for the current study, Colour, test, odour, temperature, depth of tube wells, EC (Electrical Conductivity) using conductivity meter, pH using

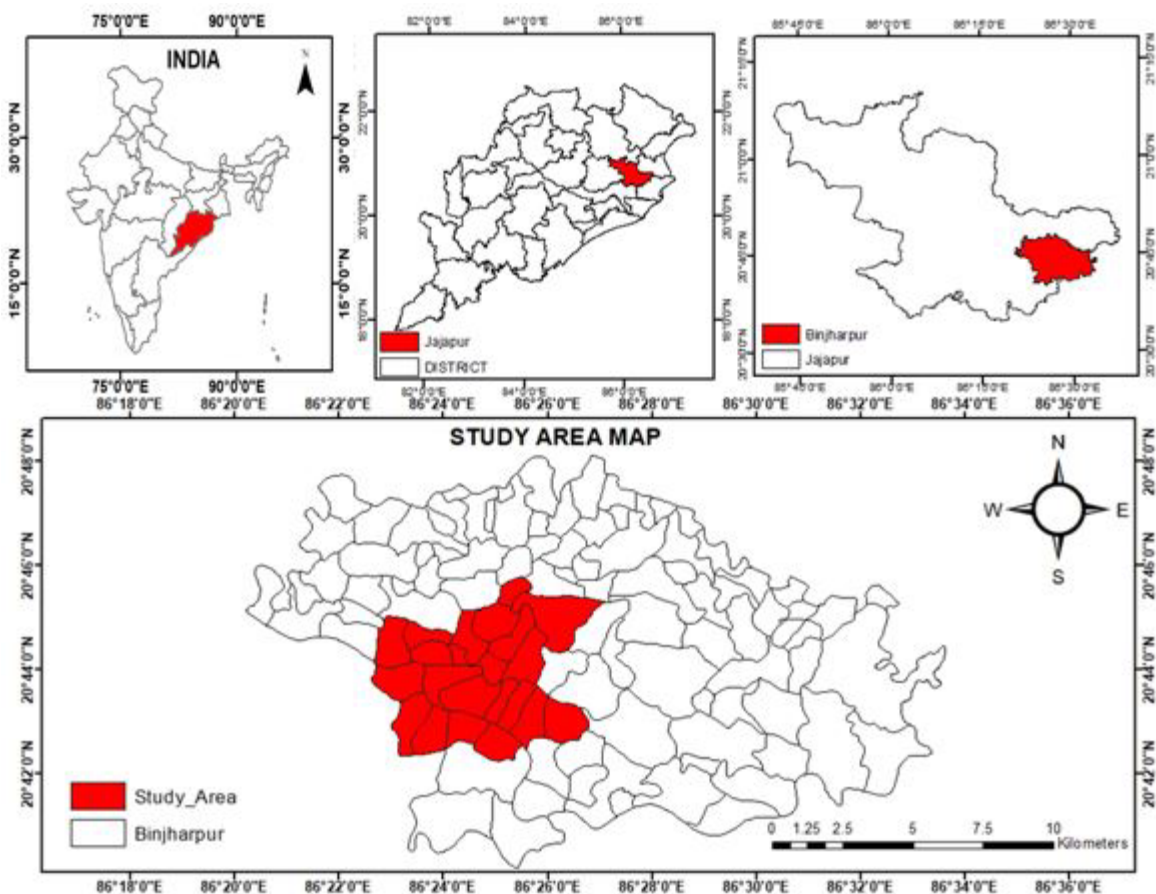


Fig. 1. Location map of the study area

pH meter and TDS (Total Dissolved Solids) using a water analyser kit (Systronics 371) were all documented. GPS readings were also recorded for each sample locations and altitudes. Local geology and rock types are also observed in the study area.  $Mg^{2+}$ ,  $Ca^{2+}$ ,  $Cl^-$ ,  $CO_3^{2-}$ ,  $HCO_3^-$ , and  $SO_4^{2-}$  were analysed by standard titration methods.  $K^+$  and  $Na^+$  were measured using the flame photometer (Systronics 128), as per the methods (Table 1) described by the American Public Health Association (APHA, 1995). Samples are analysed within a week of sampling at the chemical laboratory of our department, Sambalpur University. Water Quality Index after Brown *et al.* (1972) is calculated easily and used for overall description of the quality of water bodies in the study area. The result is expressed in comprehensible manner such as "very good", "good", "poor" for its suitability for drinking purposes. SAR, RSC, %Na, Salinity Index, SSP, KR and MH were calculated for irrigation purposes. The characteristics of groundwater quality are compared to the Bureau of Indian Standards (BIS) for drinking water.

## Results and Discussion

The finding of the Physico-chemical test was

analysed to determine the appropriateness of groundwater in the study region for drinking and irrigation uses. By comparing the measured values of several water quality indicator guideline of the BIS 1998 for drinking and irrigation purposes in Table 1. The major ions such as  $Na^+$ ,  $K^+$ ,  $Mg^{2+}$ ,  $Ca^{2+}$ ,  $SO_4^{2-}$ ,  $Cl^-$ ,  $CO_3^{2-}$  and  $HCO_3^-$  and some major physico-chemical parameters such as TDS, EC and pH, SAR, Salinity Index, %Na, RSC, SSP, KR, MH were calculated for irrigation are discussed below.

### Groundwater Quality Analysis for Drinking

#### General Parameter Analysis

The result of physico chemical parameter analysis of the collected samples is given in Table 2.

**pH** plays an important role in quality of ground water. The pH values of the collected samples vary between 6.58 and 7.81, indicating a slightly acidic to alkalinity in nature. The pH values of all the water samples in this study are within the permissible range 6.5-8.5.

**Electrical Conductivity (EC)** measures the capacity of water to conduct electric current through it. It mainly measures the dissolved ionic substances in the water sample. In the study area, EC values

**Table 1.** Physiochemical parameter of BIS guideline

Sr No.	Category of parameter	Characteristics	Analytical method	Unit	BIS Max. Permissible limit (1998)
1.	General	pH	Electrode	6.5-8.5	
2.		EC	Conductivity meter	mS/cm	3000
3.		TDS	TDS meter	mg/l	2000
4.		Total hardness (as $CaCO_3$ )	EDTA titrimetric	mg/l	600
5.	Major cations	Sodium ( $Na^+$ )	Flame photometric	mg/l	200
6.		Potassium ( $K^+$ )	Flame photometric	mg/l	200
7.		Calcium ( $Ca^{2+}$ )	EDTA titrimetric	mg/l	200
8.		Magnesium ( $Mg^{2+}$ )	EDTA titrimetric	mg/l	100
9.	Major anions	Bicarbonates ( $HCO_3^-$ )	Titrimetric	mg/l	NA
10.		Chlorides ( $Cl^-$ )	Titrimetric	mg/l	1000
11.		Nitrates ( $NO_3^-$ )	Titrimetric	mg/l	45
12.		Sulfates ( $SO_4^{2-}$ )	Barium chloride	mg/l	400
13.		Boron (B)	Curcumin method	mg/l	-
14.	Irrigation water	Salinity		%	NA
15.		SAR		-	NA
16.		RSC		mew/l	<1.25
17.		%Na		%	<20 or 20 - 40
18.		PI		%	Class 1 or 2
19.		KI		-	<1.0
20.		MH		%	Below 50%
21.	SSP		-	NA	

(Source: Pillai, G., & Khan, I. A. (2016))

ranged from 103 to 1985  $\mu\text{s}/\text{cm}$ . The EC values of all the water samples are within the permissible limit of 3000  $\mu\text{s}/\text{cm}$ .

**Total Dissolved Solid (TDS)** is a measure of all the dissolved substances in water. In this study the TDS value ranged from 77 to 1469 mg/l. The TDS measurements of all samples are within the permissible limit of 2000 mg/l as mentioned in BIS guideline (Table 1).

**Total hardness** is chemically expressed as the total concentration of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  as milligram per liter equivalent of  $\text{CaCO}_3$ . Total hardness values resulted ranged from 133 to 466 mg/l. The total hardness measurements for all the samples are below the permissible limit of 600 mg/l recommended by the BIS for drinking water (Table 1).

**Sodium ( $\text{Na}^+$ ) and Potassium ion ( $\text{K}^+$ )** concentration value varies from 10.3 to 88.3 mg/l and 0.68 to 59 mg/l respectively. The sodium ion ( $\text{Na}^+$ ) concentration values for all the samples are below the permissible limit of 200 mg/l. Out of 20 samples, 12 samples have high potassium ( $\text{K}^+$ ) content above the permissible limit of 10 mg/l; only eight (S1, S3, S5, S11, S14 to S17) samples are below the permissible limit of 10 mg/l.

**Calcium ( $\text{Ca}^{2+}$ ) and Magnesium ion ( $\text{Mg}^{2+}$ )** concentration value varies from 31 to 127 mg/l and 9.97 to

36.23 mg/l respectively. The Calcium and Magnesium ion concentration values of all the water samples are within the permissible limit of 200 mg/l and 100 mg/l respectively as recommended by the BIS (1998) for drinking purpose.

**Bicarbonate ion ( $\text{HCO}_3^-$ )** concentration value varies from 96.5 to 436.2 mg/l.

**Chlorides** are in the value ranged from 23.59 to 127.8 mg/l which is below the permissible limit of 1000 mg/l (BIS 1998).

The **Sulphate concentration ( $\text{SO}_4^{2-}$ )** of all the samples in the study area ranges from 15 to 57 mg/l which is below the permissible limit of 400 mg/l.

### Water Quality Index (WQI) Analysis

Water quality includes chemical, physical and biological characteristics of water. The result of the water quality analysis is expressed as "very good", "good", "poor" in terms of its suitability for drinking purposes. The Bureau of Indian Standards (BIS 1998) for drinking water have been considered for the calculation of WQI. Water Quality Parameters must have standard limits as prescribed by WHO/BIS/ICMR. Water quality parameters, such as temperature, pH, turbidity, dissolved oxygen, biochemical oxygen demand, nitrates and total solids are taken into consideration for generating water quality index (WQI) (Ro<sup>o</sup>u *et al.*, 2013, I. Pi<sup>o</sup>tea *et al.*, 2013;

**Table 2.** Physicochemical analysis result

Location	pH	EC	TDS	TH	Na	K	Ca	Mg	$\text{HCO}_3$	Cl	$\text{SO}_4$
S1	7.66	376	286	183	26.1	6	45	17.01	326	49.7	32
S2	7.57	875	657	157	19.3	13.7	43	11.9	436.2	34.7	37
S3	7.44	546	410	163	41.5	7.6	31	20.7	378.7	51.7	28
S4	7.38	103	77	211	34.6	12.4	65	11.7	202.4	26.81	20
S5	7.52	715	523	222	19.9	6.93	58	18.7	96.5	40.02	34
S6	7.39	224	166	174	14.3	11.6	41	17.46	126.6	36.8	28
S7	7.53	176	131	166	10.3	27	36	18.62	163.2	47.42	15
S8	7.36	701	515	133	39	10.3	35	11.1	117.3	93.91	15
S9	7.22	566	425	209	27	15	50	20.4	205.9	78.52	46
S10	6.97	907	681	208	88.3	59	53	18.37	347.2	127.8	57
S11	6.89	1985	1469	303	55	7.1	91	18.4	294.6	117.32	41
S12	7.81	852	631	263	76	44	83	13.53	199.8	74.7	54
S13	7.64	235	177	159	26.3	15.4	47	9.97	135	29.92	41
S14	6.58	792	594	169	29	0.68	40	16.91	135	36.82	24
S15	7.81	989	742	169	71	3.3	51	10.05	139.7	40.61	25
S16	7.49	901	667	157	59.2	2.4	37	15.51	128	23.59	42
S17	7.63	788	591	466	25	4.6	127	36.23	269.7	89.62	50
S18	7.36	695	522	272	56.7	30.1	83	15.61	193.9	102.8	33
S19	7.41	889	667	340	48.7	36.4	104	19.61	275.6	89.5	51
S20	7.69	694	507	281	56	21.1	83	17.93	116.7	83.9	47

Iticescu *et al.*, 2013). WQI is calculated as per the “weighted arithmetic index method” (Brown *et al.*, 1972).

$$WQI = \frac{\sum W_n Q_n}{\sum W_n}$$

Where,

$W_n$  = unity wight factor of  $n^{\text{th}}$  water quality parameters and

$Q_n$  = sub-index for water quality parameters

From the Water Quality Index, water types can be classified into different categories. (Table 3).

In the study area the WQI ranges from 31.79 to 369.33 meq/l. Out of 20 groundwater samples, 13 samples represent good quality water, 6 samples indicate poor quality water, and 1 sample indicates water as unfit for consumptions.

### Groundwater Quality Analysis for Irrigation

The quality of groundwater is also analysed for irri-

gation purposes. Calculation of SAR, KR, and MH, salinity Index, % Na, SSP, and RSC in Table 4 are the basic parameters for evaluating water quality for irrigation purposes.

### Salinity Index

Salinity index are classified on the basis of EC values which are shown in (Table 5). All the collected samples during pre-monsoon of 2018 are classified from low to high salinity class of water. 45% samples belong to high salinity category. High salinity (class 3) in the study area shows that the water is suitable for irrigation purposes.

### Sodium Absorption Ratio (SAR)

The Sodium Absorption Ratio is an important water quality parameter for measuring suitability of water for agricultural purposes. (Reeve *et al.*, 1954).

The formula for calculating the sodium adsorption ratio (SAR) is

**Table 3.** Classification of Water based on WQI

WQI developed by Brown *et al.*, (1972)

Water Quality Index	Water Quality	No of samples
0-25	Excellent	-
26-50	Good	13
51-75	Poor	6
76-100	Very Poor	
>100	Unfit for consumption	1

**Table 4.** Groundwater quality value for irrigation

Location	SAR	%Na	SSP	RSC	KI	MH
S1	4.69	34.10	29.62206333	263.99	0.420899855	0.000222222
S2	3.68	37.54	26.01078167	381.3	0.35154827	0.000232558
S3	8.16	48.71	44.527897	327	0.80270793	0.000322581
S4	5.59	37.99	31.08715184	125.7	0.451108214	0.000153846
S5	3.21	25.91	20.60041408	19.8	0.259452412	0.000172414
S6	2.65	30.70	19.65365585	68.14	0.2446117	0.000243902
S7	1.97	40.57	15.86568084	108.58	0.188575613	0.000277778
S8	8.13	51.67	45.82843713	71.2	0.845986985	0.000285714
S9	7.58	37.36	27.72073922	135.5	0.383522727	0.0002
S10	14.8	67.36	55.30155947	275.83	1.237214516	0.000188679
S11	7.31	36.20	33.45498783	185.2	0.50274223	0.00010989
S12	10.9	55.41	44.05031009	103.27	0.787320004	0.000120482
S13	7.6	42.26	31.58400384	78.03	0.461646481	0.000212766
S14	5.61	34.27	33.75625655	78.09	0.509576524	0.00025
S15	12.9	54.89	53.76751231	78.65	1.162981163	0.000196078
S16	11.6	53.98	52.9943604	75.49	1.127404304	0.00027027
S17	2.77	15.35	13.28162355	106.47	0.15315812	7.87402E-05
S18	8.08	46.81	36.5076299	95.29	0.574992394	0.000120482
S19	6.21	40.77	28.26301433	151.99	0.393981069	9.61538E-05
S20	4.96	43.30	35.68470018	15.77	0.554839988	0.000120482

$$\text{SAR} = \frac{\text{Na}^+}{\left\{ \frac{\text{Ca}^{2+} + \text{Mg}^{2+}}{2} \right\}^{0.5}}$$

Where Sodium, Calcium and Magnesium concentration are expressed in meq/l.

The SAR values of the collected groundwater samples from the study area are given in (Table 6). During pre-monsoon period, the SAR value of all the samples ranges from 1.97 to 14.78 meq/l. SAR value of 16 samples are found to be below 10 meq/l which are classified as excellent for irrigation purpose (class S1).

#### 4.2.3. %Na

Percent sodium plays an important role to assess the suitability of water for irrigation purpose (Wilcox, 1948). The result of percent sodium of groundwater samples from the study area are given in Table 7. %Na value of all the samples varies from 15.35 to 67.36 mg/l. One samples has an excellent water classes, eight samples have good water class, ten

samples have a permissible water class and just one sample has a doubtful water class.

$$\% \text{Na} = \left( \frac{\text{Na}^+ + \text{K}^+}{(\text{Ca}^{2+} + \text{Mg}^{2+} + \text{Na}^+ + \text{K}^+)} \right) * 100$$

#### Soluble Sodium Percentage (SSP)

The Soluble Sodium Percentage aids in permeability classification of irrigation water.  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  ions found in clay particles have a tendency to exchange Sodium ions contained in irrigation water. This lowers the permeability of the material, lowering its quality. The proportion of soluble sodium percentage in irrigation water is used to classify it (Todd 1959).

**Table 8.** Groundwater quality based on RSC.

RSC (meq/l)	Remark on quality	Samples
<1.25	Good	—
1.25-2.50	doubtful	—
>2.50	unsuitable	20

**Table 5.** Classification of water samples based on of EC.

EC ( $\mu\text{S}/\text{cm}$ )	Water salinity	Range (no. of samples)	Percent
0 - 250	Low	103-235 (4 samples)	20%
251 - 750	Medium	376-715 (7 samples)	35%
751 - 2250	High	788-1985 (9 samples)	45%
2251 - 6000	Very High	-	-
6001 - 10,000	Extensively High	-	-
10,001 - 20,000	Brines weak concentration	-	-
20,001 - 50,000	Brines moderate concentration	-	-
50,001 - 100,000	Brines high concentration	-	-
>100,000	Brines extremely high concentration	-	-

**Table 6.** Classification of waters based on SAR values.

SAR values	Sodium hazard class	Remark	Range and no. of Sample
<10	S1	Excellent	1.97 to 8.16 (16 samples)
10 - 18	S2	Good	10.93 to 14.78 (4 samples)
19 - 26	S3	Doubtful/fair poor	—
>26	S4	Unsuitable	—

**Table 7.** Percentage Sodium water class

Sodium (%)	Class	Range (No of samples)
<20	Excellent	15.35 (1sample)
20 - 40	Good	25.91 to 37.99 (8 samples)
40 - 60	Permissible	4.57 to 55.41 (10 samples)
60 - 80	Doubtful	67.36 (1 sample)
>80	Unsuitable	—

**Table 9.** Kelly's ratio

	KR/K%	
Groundwater samples	<1	>1
	Suitable	Unsuitable
Pre-monsoon	17 samples	3 samples

$$SSP = \left( \frac{Na^+}{(Ca^{2+} + Mg^{2+} + Na^+)} \right) * 100$$

The soluble sodium percentage values of the water samples in this study are from 13.28 to 55.30 meq/L

### Residual Sodium Carbonate (RSC)

The suitability of groundwater for irrigation is also affected by the excess amount of Carbonate and Bicarbonate concentration in comparison to the Calcium and Magnesium ions present in water (Richards, 1954). The excess Bicarbonate and Carbonate is alluded to as Residual Sodium Carbonate.

$$RSC = (CO_3 + HCO_3) - (Ca + Mg)$$

The RSC value of the collected samples ranges from 15.77 to 381.3 meq/l (Table 8). Based on the RSC values <2.50 meq/l, all the samples are classified as unsuitable for irrigation purposes.

### Kelly's Index (KI)

Kelly (1940) introduced a factor to assess the suitability of water for irrigation purpose using the concentration of Sodium against Calcium and Magnesium (Table 9).

$$\text{Kelly's ratio} = Na^+ / (Ca^{2+} + Mg^{2+})$$

In this study, Kelly's ratio varies from 0.15 to 1.23 meq/. Only three samples (S10, S15 and S16) are found unsuitable for irrigation purpose.

### Conclusion

In the present study in parts of Jajpur district, the result of physicochemical parameters analysis of groundwater samples is within the permissible limit as compared to Bureau of Indian Standards (BIS 1998) for drinking purpose. WQI value of all the samples ranges from 31.79 to 369.33 meq/l. 13 samples represent good quality water, 6 samples indicate poor quality water, and only 1 sample indicates water to be unfit for consumptions for drinking purpose. Out of the various indices calculated for the suitability of water for irrigation purpose, most of the samples are found suitable for irrigation purpose whereas very few of them are found unfit for irrigation purpose.

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### References

- Aryal, J., Gautam, B. and Sapkota, N. 2012. Drinking water quality assessment. *Journal of Nepal Health Research Council*. 10(22) : 192-196.
- Ayers, R. S. 1977. Quality of water for irrigation. *Journal of the irrigation and Drainage Division*. 103(2): 135-154.
- Breabăn, I. G., Ghepeu, D. and Paiu, M. 2012. Determination of Water Quality Index of Jijia and Miletin Ponds Iuliana Gabriela BREABĂN1), Diana GHEPEU2), Mădălina PAIU1). *Bulletin UASVM Agriculture*. 69: 2.
- Christiansen, J. E., Olsen, E. C. and Willardson, L. S. 1977. Irrigation water quality evaluation. *Journal of the Irrigation and Drainage Division*. 103(2) : 155-169.
- Horneck, D. A., Ellsworth, J. W., Hopkins, B. G., Sullivan, D. M. and Stevens, R. G. 2007. Managing salt-affected soils for crop production.
- Kelly, W. P. 1940. Permissible composition and concentration of irrigated waters. *Proceedings of the ASCF66*. 607.
- Mohamed, A. K., Dan, L., Kai, S., Eldaw, E. and Abualela, S. 2019. Evaluating the suitability of groundwater for drinking purposes in the North Chengdu Plain, China. In *E3S Web of Conferences* (Vol. 81, p. 01006). EDP Sciences.
- Pillai, G. and Khan, I. A. 2016. Assessment of groundwater suitability for drinking and irrigation purpose in the Dimbhe Command Area of River Ghod, Maharashtra, India. *Journal of Geoscience and Environment Protection*. 4(12): 142-157.
- Pistea, I., Rosu, C., Martonos, I. and Ozunu, A. 2013. Romanian surface water quality: Tarnava Mare River between Medias and Copsa Mica case study. *Environmental Engineering and Management Journal*. 12(2): 283-289.
- Reeve, R. C., Bower, C. A., Brooks, R. H. and Gschwend, F. B. 1954. A comparison of the effects of exchangeable sodium and potassium upon the physical condition of soils. *Soil Science Society of America Journal*. 18(2) : 130-132.
- Rosu, C., Pistea, I., Calugar, M., Martonos, I. and Ozunu, A. 2013. Assessment of ground water quality status by using water quality index (WQI) method in Tureni village, Cluj County. *Aerul si Apa. Componente ale Mediului*. 111.
- Standard, B. I. S. 1998. Specifications for drinking water. *Bureau of Indian Standard*. 171-178.
- Todd, D.K. 1959. *Groundwater Hydrology*. Wiley, New York, 535.
- Wilcox, L. V. 1958. *Determining the quality of irrigation water* (No. 197). US Department of Agriculture.